

Al-Nd-Si (Aluminum-Neodymium-Silicon)

V. Raghavan

Recently, [2001Lon1] and [2001Lon2] reported an isothermal section at 500 °C for this ternary system in the Nd-lean region.

Binary Systems

The Al-Nd phase diagram, reassessed with new additional experimental input by [2005Gao], shows the following intermediate compounds: Nd₃Al (*D*0₁₉, Ni₃Sn-type hexagonal), Nd₂Al (*C*23, Co₂Si-type orthorhombic), NdAl (ErAl-type orthorhombic), NdAl₂ (*C*15, MgCu₂-type cubic), αNdAl₃ (Ni₃Sn-type hexagonal), βNdAl₃ (stable between 1205 and 888 °C), NdAl₄ or βNd₃Al₁₁ (*D*1₃, Al₄Ba-type tetragonal), and αNd₃Al₁₁ (αLa₃Al₁₁-type orthorhombic). The Al-Si phase diagram is a simple eutectic system with the eutectic reaction at 577 °C and 12.2 at.% Si. The Nd-Si phase diagram [Massalski2] depicts the following intermediate phases: Nd₅Si₃ (*D*8_l, Cr₅Si₃-type tetragonal), Nd₅Si₄ (Si₄Zr₅-type tetragonal), NdSi (*B*27, FeB-type orthorhombic), Nd₂Si₃ (*C*32, AlB₂-type hexagonal), αNdSi₂ (αGdSi₂-type orthorhombic), and βNdSi₂ (*C*_c, ThSi₂-type tetragonal).

Ternary Isothermal Section

With starting metals of 99.99% Al, 99.9% Nd, and 99.999% Si, [2001Lon2] arc-melted or induction-melted 260 alloys under Ar atm. After homogenization, the alloys were given a final anneal at 500 °C for 7 days and quenched in water. The phase equilibria were studied by metallography, electron probe microanalysis, and x-ray powder diffraction. The isothermal section constructed by

[2001Lon2] at 500 °C is shown in Fig. 1. The ternary compound labeled δ by [2001Lon2] at NdAl_{1.75}Si_{0.25} is an extension of the binary compound NdAl₂ with the MgCu₂-type cubic structure [1967Ram]. The compound NdAl_{1.25}Si_{0.75} (or Nd₄Al₅Si₃, denoted τ₁ here) has a homogeneity range of NdAl_{1+x}Si_{1-x} (0 ≤ x ≤ 0.4) [2001Lon1]. The compound of fixed stoichiometry NdAl₂Si₂ (denoted τ₂ here) has the La₂O₂S-type of structure with space group *P*3̄*m*1 and lattice parameters *a* = 0.423 nm and *c* = 0.671 nm [1984Gsc].

References

- 1967Ram:** A. Raman and H. Steinfink, Crystal Chemistry of AB₂ Structures. I. Investigations on AB₂ Sections in the Ternary Systems RE-Al-Si, -Ge, and -Sn, *Inorg. Chem.*, 1967, **6**(10), p 1789-1791
- 1984Gsc:** K.A. Gschneidner and L. Eyring, *Handbook on the Physics and Chemistry of Rare Earths*, 1984, Vol. 7, North-Holland, Amsterdam, p 140
- 2001Lon1:** Z. Long, Y. Zhou, Y. Zhuang, R. Chen, and J. Liu, Isothermal Section (500 °C) of Phase Diagram of Nd-Al-Si Ternary System, *Trans. Nonferrous Met. Soc. China*, 2001, **11**(2), p 278-280
- 2001Lon2:** Z. Long, Y. Zhou, Y. Zhuang, R. Chen, J. Liu, and X. Wang, Phase Relations in the Nd-Al-Si System at 500 °C, *J. Alloys Compd.*, 2001, **325**, p 190-193
- 2005Gao:** M.C. Gao, N. Unlu, G.J. Shiflet, M. Mihalkovic, and M. Widom, Reassessment of Al-Ce and Al-Nd Binary Systems Supported by Critical Experiments and First-Principles Energy Calculation, *Metall. Mater. Trans. A*, 2005, **36**, p 3269-3279

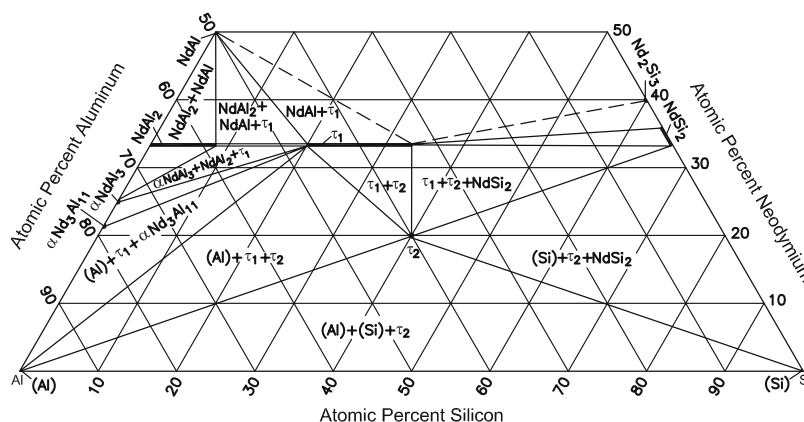


Fig. 1 Al-Nd-Si isothermal section at 500 °C [2001Lon2]. Narrow two-phase regions are omitted